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This volume is a comparative study of the practice of impagination across different ages and civilizations. By impagination we mean the act of placing and arranging spatially textual and other information onto a material bearer that could be made of a variety of materials (papyrus, bamboo slips, palm leaf, parchment, paper, and the computer screen). This volume investigates three levels of impagination: what is the page or other unit of the material bearer,

what is written or printed on it, and how is writing or print placed on it. It also examines the interrelations of two or all three of these levels. Collectively it examines the material and materiality of the page, the variety of imprints, cultural and historical conventions for impagination, interlinguistic encounters, the control of editors, scribes, publishers and readers over the page, inheritance, borrowing and innovation, economics, aesthetics and socialities of imprints and impagination, and the

relationship of impagination to philology. This volume supplements studies on mise en page and layout – an important subject of codicology – first by including non-codex writings, second by taking a closer look at the page or other unit than at the codex (or book), and third by its aspiration to adopt a globally comparative approach. This volume brings together for comparison vast geographical realms of learning, including Europe, China, Tibet, Korea, Japan and the Near Eastern and European

communities in which the Hebrew Bible was transmitted. This comparison is significant, for Europe, China, and India all developed great traditions of learning which came into intensive contact. The contributions to this volume are firmly rooted in local cultures and together address global, comparative themes that are significant for multiple disciplines, such as intellectual and cultural history of knowledge (both humanistic and scientific), global history, literary and media studies, aesthetics, and studies of material culture, among other fields.

"The fourteenth edition continues a long tradition of providing a firm foundation in the concepts of chemical principles while instilling an appreciation of the important role chemistry plays in our daily lives. We believe that it is our responsibility to assist both instructors and students in their pursuit of this goal by presenting a broad range of chemical topics in a logical format. At all times, we strive to balance theory and application and to illustrate principles with applicable examples whenever possible"--

Colloid and Interface Chemistry for Water Quality Control provides basic but essen-

tial knowledge of colloid and interface science for water and wastewater treatment. Divided into two sections, chapters 1 to 8 presents colloid chemistry including simple history and basic concepts, diffusion and Brown Motion, sedimentation, osmotic pressure, optical properties, rheology properties, electric properties, emulsion, foam and gel, and so on; chapters 9 to provides interface chemistry theories including the surface of liquid, the surface of solution, and the surface of solid. This valuable book is the only one that presents colloid and interface chemistry from the water quality control perspective. This book was written for graduate students in the area of water treatment and environmental engineering, and it could be used as the reference for researchers and engineers in the same area. Concise content makes this suitable for both teaching and learning. Focuses on water treatment technology and methods, links colloid and surface chemistry to water treatment applications. Not only addresses all the important physical-chemistry principles and theories, but also presents new developed knowledge on water treatment. Includes exercises, problems and solutions, which are very

helpful for testing learning and understanding

This unique compendium gives an updated presentation of clustering, one of the most challenging tasks in machine learning. The book provides a unitary presentation of classical and contemporary algorithms ranging from partitional and hierarchical clustering up to density-based clustering, clustering of categorical data, and spectral clustering. Most of the mathematical background is provided in appendices, highlighting algebraic and complexity theory, in order to make this volume as self-contained as possible. A substantial number of exercises and supplements makes this a useful reference textbook for researchers and students.

Cellulose Nanoparticles: Chemistry and Fundamentals covers the synthesis, characterization and processing of cellulose nanomaterials.

Essentials of Computational Chemistry provides a balanced introduction to this dynamic subject. Suitable for both experimentalists and theorists, a wide range of samples and applications are included drawn from all key areas. The book careful-

ly leads the reader through the necessary equations providing information explanations and reasoning where necessary and firmly placing each equation in context.

INTERNATIONAL BESTSELLER "For anyone who wants to understand capitalism not as economists or politicians have pictured it but as it actually operates, this book will be invaluable."-Observer (UK) If you've wondered how we did not see the economic collapse coming, Ha-Joon Chang knows the answer: We didn't ask what they didn't tell us about capitalism. This is a lighthearted book with a serious purpose: to question the assumptions behind the dogma and sheer hype that the dominant school of neoliberal economists-the apostles of the freemarket-have spun since the Age of Reagan. Chang, the author of the international bestseller *Bad Samaritans*, is one of the world's most respected economists, a voice of sanity-and wit-in the tradition of John Kenneth Galbraith and Joseph Stiglitz. *23 Things They Don't Tell You About Capitalism* equips readers with an understanding of how global capitalism works-and doesn't. In his final chapter, "How to Rebuild the World," Chang offers a vision of how we can shape capitalism to humane

ends, instead of becoming slaves of the market.

The two-part, fifth edition of *Advanced Organic Chemistry* has been substantially revised and reorganized for greater clarity. The material has been updated to reflect advances in the field since the previous edition, especially in computational chemistry. Part A covers fundamental structural topics and basic mechanistic types. It can stand-alone; together, with Part B: *Reaction and Synthesis*, the two volumes provide a comprehensive foundation for the study in organic chemistry. Companion websites provide digital models for study of structure, reaction and selectivity for students and exercise solutions for instructors.

One of the best-selling books for AP Chemistry, this 11th AP Edition continues the tradition of excellence. Chemistry features a straightforward writing style and proven problem-solving strategies that make this text ideal for the AP Chemistry classroom. In this edition students will be guided by the chapter opener Essential Questions that pinpoint the essential AP content that they will study in the chapter. Additionally,

each chapter ends with a Look Back at the AP Essential Knowledge reviewing the most important chapter concepts. In addition to these tools this AP Edition also includes Chapter Openers that summarize how the Big Ideas are covered in the chapter.

Since the discovery of the first examples of 2-oxoglutarate-dependent oxygenase-catalysed reactions in the 1960s, a remarkably broad diversity of alternate reactions and substrates has been revealed, and extensive advances have been achieved in our understanding of the structures and catalytic mechanisms. These enzymes are important agrochemical targets and are being pursued as therapeutic targets for a wide range of diseases including cancer and anemia. This book provides a central source of information that summarizes the key features of the essential group of 2-oxoglutarate-dependent dioxygenases and related enzymes. Given the numerous recent advances and biomedical interest in the field, this book aims to unite the latest research for those already working in the field as well as to provide an introduction for those newly approaching the topic, and for those interested in translating the ba-

sic science into medicinal and agricultural benefits. The book begins with four broad chapters that highlight critical aspects, including an overview of possible catalytic reactions, structures and mechanisms. The following seventeen chapters focus on carefully selected topics, each written by leading experts in the area. Readers will find explanations of rapidly evolving research, from the chemistry of isopenicillin N synthase to the oxidation mechanism of 5-methylcytosine in DNA by ten-eleven--translocase oxygenases.

Chang's best-selling general chemistry textbook takes a traditional approach and is often considered a student and teacher favorite. The book features a straightforward, clear writing style and proven problem-solving strategies. It continues the tradition of providing a firm foundation in chemical concepts and principles while presenting a broad range of topics in a clear, concise manner. The tradition of "Chemistry" has a new addition with co-author, Kenneth Goldsby from Florida State University, adding variations to the 12th edition. The organization of the chapter order has changed with nuclear chemistry moving up in the chapter order.

The principles of general chemistry, stressing the underlying concepts in chemistry, relating abstract concepts to specific real-world examples, and providing a programme of problem-solving pedagogy.

Atkins' Physical Chemistry: Molecular Thermodynamics and Kinetics is designed for use on the second semester of a quantum-first physical chemistry course. Based on the hugely popular Atkins' Physical Chemistry, this volume approaches molecular thermodynamics with the assumption that students will have studied quantum mechanics in their first semester. The exceptional quality of previous editions has been built upon to make this new edition of Atkins' Physical Chemistry even more closely suited to the needs of both lecturers and students. Re-organised into discrete 'topics', the text is more flexible to teach from and more readable for students. Now in its eleventh edition, the text has been enhanced with additional learning features and maths support to demonstrate the absolute centrality of mathematics to physical chemistry. Increasing the digestibility of the text in this new approach, the reader is brought to a question, then

the math is used to show how it can be answered and progress made. The expanded and redistributed maths support also includes new 'Chemist's toolkits' which provide students with succinct reminders of mathematical concepts and techniques right where they need them. Checklists of key concepts at the end of each topic add to the extensive learning support provided throughout the book, to reinforce the main take-home messages in each section. The coupling of the broad coverage of the subject with a structure and use of pedagogy that is even more innovative will ensure Atkins' Physical Chemistry remains the textbook of choice for studying physical chemistry.

Providing vital knowledge on the design and synthesis of specific metal-organic framework (MOF) classes as well as their properties, this ready reference summarizes the state of the art in chemistry. Divided into four parts, the first begins with a basic introduction to typical cluster units or coordination geometries and provides examples of recent and advanced MOF structures and applications typical for the respective class. Part II covers recent progress in linker chemistries, while spe-

cial MOF classes and morphology design are described in Part III. The fourth part deals with advanced characterization techniques, such as NMR, in situ studies, and modelling. A final unique feature is the inclusion of data sheets of commercially available MOFs in the appendix, enabling experts and newcomers to the field to select the appropriate MOF for a desired application. A must-have reference for chemists, materials scientists, and engineers in academia and industry working in the field of catalysis, gas and water purification, energy storage, separation, and sensors.

Tensors have numerous applications in physics and engineering. There is often a fuzzy haze surrounding the concept of tensor that puzzles many students. The old-fashioned definition is difficult to understand because it is not rigorous; the modern definitions are difficult to understand because they are rigorous but at a cost of being more abstract and less intuitive. The goal of this book is to elucidate the concepts in an intuitive way but without loss of rigor, to help students gain deeper understanding. As a result, they will not need to recite those definitions in a parrot-like

manner any more. This volume answers common questions and corrects many misconceptions about tensors. A large number of illuminating illustrations helps the reader to understand the concepts more easily. This unique reference text will benefit researchers, professionals, academics, graduate students and undergraduate students.

The seventh edition of General Chemistry continues the tradition of presenting only the material that is essential for a one-year general chemistry course. It strikes a balance between theory and application by incorporating real-world examples; helping students visualize the three-dimensional atomic and molecular structures that are the basis of chemical activity; and developing problem-solving and critical thinking skills. Although the seventh edition incorporates many impressive features, such as conceptual idea review, animations correlated to the text, and hand-drawn worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr. Chang and Dr. Goldsby's concise-but-thorough approach will appeal to efficiency-minded instructors

and value-conscious students.

Following in the wake of Chang's two other best-selling physical chemistry textbooks (Physical Chemistry for the Chemical and Biological Sciences and Physical Chemistry for the Biosciences), this new title introduces laser spectroscopist Jay Thoman (Williams College) as co-author. This comprehensive new text has been extensively revised both in level and scope. Targeted to a mainstream physical chemistry course, this text features extensively revised chapters on quantum mechanics and spectroscopy, many new chapter-ending problems, and updated references, while biological topics have been largely relegated to the previous two textbooks. Other topics added include the law of corresponding states, the Joule-Thomson effect, the meaning of entropy, multiple equilibria and coupled reactions, and chemiluminescence and bioluminescence. One way to gauge the level of this new text is that students who have used it will be well prepared for their GRE exams in the subject. Careful pedagogy and clear writing throughout combine to make this an excellent choice for your physical chemistry

course.

The Study Guide includes learning goals, an overview, a review section with worked examples, and self-tests with answers.

One of the most cited books in physics of all time, Quantum Computation and Quantum Information remains the best textbook in this exciting field of science. This 10th anniversary edition includes an introduction from the authors setting the work in context. This comprehensive textbook describes such remarkable effects as fast quantum algorithms, quantum teleportation, quantum cryptography and quantum error-correction. Quantum mechanics and computer science are introduced before moving on to describe what a quantum computer is, how it can be used to solve problems faster than 'classical' computers and its real-world implementation. It concludes with an in-depth treatment of quantum information. Containing a wealth of figures and exercises, this well-known textbook is ideal for courses on the subject, and will interest beginning graduate students and researchers in physics, computer science, mathematics, and electrical engineering.

The Student Solutions Manual will have all

the solutions to the even numbered problems in the text. The style of the solutions will match worked examples in the text to help the student learn how to solve the problems.

The Sixth Edition of a classic in organic chemistry continues its tradition of excellence. Now in its sixth edition, March's Advanced Organic Chemistry remains the gold standard in organic chemistry. Throughout its six editions, students and chemists from around the world have relied on it as an essential resource for planning and executing synthetic reactions. The Sixth Edition brings the text completely current with the most recent organic reactions. In addition, the references have been updated to enable readers to find the latest primary and review literature with ease. New features include: More than 25,000 references to the literature to facilitate further research. Revised mechanisms, where required, that explain concepts in clear modern terms. Revisions and updates to each chapter to bring them all fully up to date with the latest reactions and discoveries. A revised Appendix B to facilitate correlating chapter sections with

synthetic transformations

Aimed at the one-year general chemistry course, this text offers a shorter, more compact presentation of topics at the same depth and with the same rigor as other traditional mainstream texts. It includes only the core topics necessary for a good foundation in general chemistry but without sacrificing clarity and comprehension.

The new edition of this best-selling general chemistry text continues to provide a firm foundation in chemical concepts and principles, while presenting a broad range of topics in a concise manner. A hallmark of this edition is the integration of many tools designed to inspire both students and instructors.

Aimed at students and professionals, this book covers every major aspect of petroleum: the origin of fossil hydrocarbons and their chemical/physical properties; discovering hydrocarbon reserves; recovering oil, gas, and bitumen; purifying gas; the chemical and physical characterization of crude oil; refining crudes into fuels and lubricants; and converting simple chemicals into solvents, polymers, fibers, rubbers, coatings, and myriad other prod-

ucts, including pharmaceuticals. Readers will learn how the industry operates, from "upstream" exploration and production, "midstream" transportation to "downstream" refining, and manufacturing of finished products. The book also contains unique chapters on midstream operations, learnings from major accidents, and safety/environmental laws and regulations. It builds on the authors' previous books and teaching material from a highly rated course that is taught at the Florida A&M University/Florida State University (USA).

Human chemistry is the study of bond-forming and bond-breaking reactions between people and the structures they form. People often speak of having either good or bad chemistry together: whereby, according to consensus, the phenomenon of love is a chemical reaction. The new science of human chemistry is the study of these reactions. Historically, human chemistry was founded with the 1809 publication of the classic novella *Elective Affinities*, by German polymath Johann von Goethe, a chemical treatise on the origin of love. Goethe based his human chemistry on Swedish chemist Torbern Berg-

man's 1775 chemistry textbook *A Dissertation on Elective Attractions*, which itself was founded on Isaac Newton's 1687 supposition that the cause of chemical phenomena may 'all depend upon certain forces by which the particles of bodies, by some causes hitherto unknown, are either mutually impelled towards each other, and cohere in regular figures, or are repelled and recede from one another'; which thus defines life.

Must-have reference for processes involving liquids, gases, and mixtures Reap the time-saving, mistake-avoiding benefits enjoyed by thousands of chemical and process design engineers, research scientists, and educators. *Properties of Gases and Liquids*, Fifth Edition, is an all-inclusive, critical survey of the most reliable estimating methods in use today --now completely rewritten and reorganized by Bruce Poling, John Prausnitz, and John O'Connell to reflect every late-breaking development. You get on-the-spot information for estimating both physical and thermodynamic properties in the absence of experimental data with this property data bank of 600+ compound constants. Bridge the gap between theory and practice with this trust-

ed, irreplaceable, and expert-authored expert guide -- the only book that includes a critical analysis of existing methods as well as hands-on practical recommendations. Areas covered include pure component constants; thermodynamic properties of ideal gases, pure components and mixtures; pressure-volume-temperature relationships; vapor pressures and enthalpies of vaporization of pure fluids; fluid phase equilibria in multicomponent systems; viscosity; thermal conductivity; diffusion coefficients; and surface tension.

Filling the need for an up-to-date handbook, this ready reference closely investigates the use of CO₂ for ureas, enzymes, carbamates, and isocyanates, as well as its use as a solvent, in electrochemistry, biomass utilization and much more. Edited by an internationally renowned and experienced researcher, this is a comprehensive source for every synthetic chemist in academia and industry.

Chang's best-selling general chemistry textbook takes a traditional approach and is often considered a student and teacher favorite. The book features a straightforward, clear writing style and proven problem-solving strategies. It continues

the tradition of providing a firm foundation in chemical concepts and principles while presenting a broad range of topics in a clear, concise manner.

Physical Chemistry for the Biosciences has been optimized for a one-semester introductory course in physical chemistry for students of biosciences.

Hailed by advance reviewers as "a kinder, gentler P. Chem. text," this book meets the needs of an introductory course on physical chemistry, and is an ideal choice for courses geared toward pre-medical and life sciences students. Physical Chemistry for the Chemical and Biological Sciences offers a wealth of applications to biological problems, numerous worked examples and around 1000 chapter-end problems.

The first IUPAC Manual of Symbols and Terminology for Physicochemical Quantities and Units (the Green Book) of which this is the direct successor, was published in 1969, with the object of 'securing clarity and precision, and wider agreement in the use of symbols, by chemists in different countries, among physicists, chemists and engineers, and by editors of scientific jour-

nals'. Subsequent revisions have taken account of many developments in the field, culminating in the major extension and revision represented by the 1988 edition under the simplified title Quantities, Units and Symbols in Physical Chemistry. This 2007, Third Edition, is a further revision of the material which reflects the experience of the contributors with the previous editions. The book has been systematically brought up to date and new sections have been added. It strives to improve the exchange of scientific information among the readers in different disciplines and across different nations. In a rapidly expanding volume of scientific literature where each discipline has a tendency to retreat into its own jargon this book attempts to provide a readable compilation of widely used terms and symbols from many sources together with brief understandable definitions. This is the definitive guide for scientists and organizations working across a multitude of disciplines requiring internationally approved nomenclature.

To understand, maintain, and protect the physical environment, a basic understanding of chemistry, biology, and physics, and

their hybrids is useful. Rapid Review of Chemistry for the Life Sciences and Engineering demystifies chemistry for the non-chemist who, nevertheless, may be a practitioner of some area of science or engineering requiring or involving chemistry. It provides quick and easy access to fundamental chemical principles, quantitative relationships, and formulas. Armed with select, contemporary applications, it is written in the hope to bridge a gap between chemists and non-chemists, so that they may communicate with and understand each other. Chapters 1-10 are designed to contain the standard material in an introductory college chemistry course. Chapters 11-15 present applications of chemistry that should interest and appeal to scientists and engineers engaged in a variety of fields. Additional features More than 100 solved examples clearly illustrated and explained with SI units and conversion to other units using conversion tables included Assists the reader to understand organic and inorganic compounds along with their structures, including isomers, enantiomers, and congeners of organic compounds Provides a quick and easy access to basic chemical concepts and spe-

cific examples of solved problems. Ideal sidekick for students who are non-chemistry majors taking intro. college chemistry, needing clear, concise explanations. This concise, user-friendly review of general and organic chemistry with environmental applications will be of interest to all disciplines and backgrounds.

Designed for the two-semester general chemistry course, Chang's best-selling textbook continues to take a traditional approach and is often considered a student and teacher favorite. The book features a straightforward, clear writing style and

proven problem-solving strategies. It continues the tradition of providing a firm foundation in chemical concepts and principles while presenting a broad range of topics in a clear, concise manner. The tradition of "Chemistry" has a new addition with co-author, Kenneth Goldsby from Florida State University, adding variations to the 11th edition. The organization of the chapter order has changed with nuclear chemistry moving up in the chapter order. There is a new problem type - Interpreting, Modeling, and Estimating - fully demonstrating what a real life chemist does on a daily basis.

The authors have added over 340 new problems to the book. The new edition of "Chemistry" continues to strike a balance between theory and application by incorporating real examples and helping students visualize the three-dimensional atomic and molecular structures that are the basis of chemical activity. An integral part of the text is to develop students' problem-solving and critical thinking skills. The 11th edition continues to deliver the integration of tools designed to inspire both students and instructors. Effective technology is integrated throughout the book.