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Write correct formulas for the products in these double replacement reactions. $\text{H}_3\text{PO}_4 + 2\text{K}_2\text{CO}_3 + \text{BaCl}_2 \rightarrow \text{Ba}_3(\text{PO}_4)_2 + 6\text{KCl}$
 $\text{Co} + 5\text{AgNO}_3 + \text{KCl} \rightarrow \text{CoCl}_2 + 5\text{KNO}_3 + \text{Ag}$
 $\text{H}_2\text{SO}_4 + \text{CaO} \rightarrow \text{CaSO}_4 + \text{H}_2\text{O}$
 $\text{Ag}_2\text{CO}_3 + \text{K}_2\text{CrO}_4 \rightarrow \text{K}_2\text{Ag}_2\text{CO}_7 + \text{CrO}_3$

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